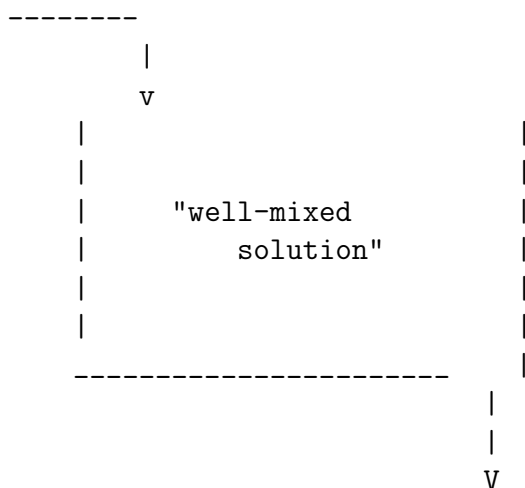


**Mixing problems in ODEs**  
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Suppose that we have two chemical substances where one is soluble in the other, such as salt and water. Suppose that we have a tank containing a mixture of these substances, and the mixture of them is poured in and the resulting “well-mixed” solution pours out through a valve at the bottom. (The term “well-mixed” is used to indicate that the fluid being poured in is assumed to instantly dissolve into a homogeneous mixture the moment it goes into the tank.) The crude picture looks like this:



Assume for concreteness that the chemical substances are salt and water. Let

- $A(t)$  denote the amount of salt at time  $t$ ,
- “flow-rate-in” = the rate at which the solution pours into the tank,
- “flow-rate-out” = the rate at which the mixture pours out of the tank,
- $C_{in}$  = “concentration in” = the concentration of salt in the solution being poured into the tank,
- $C_{out}$  = “concentration out” = the concentration of salt in the solution being poured out of the tank,

- $R_{in}$  = rate at which the salt is being poured into the tank = (“flow-rate-in”)( $C_{in}$ ),
- $R_{out}$  = rate at which the salt is being poured out of the tank = (“flow-rate-out”)( $C_{out}$ ).

Notes: (1) If flow-rate-in = flow-rate-out then the “water level” of the tank stays the same. (2) We can determine  $C_{out}$  as a function of other quantities:

$$C_{out} = \frac{A(t)}{T(t)},$$

where  $T(t)$  denotes the volume of solution in the tank at time  $t$ . (3) The rate of change of the amount of salt in the tank,  $A'(t)$ , more properly could be called the “net rate of change”. If you think of it this way then you see  $A'(t) = R_{in} - R_{out}$ .

Now the differential equation for the amount of salt arises from the above equations:

$$A'(t) = (\text{“flow - rate - in”})C_{in} - (\text{“flow - rate - out”})\frac{A(t)}{T(t)}.$$

**Example:** Consider a tank with 200 liters of salt-water solution, 30 grams of which is salt. Pouring into the tank is a brine solution at a rate of 4 liters/minute and with a concentration of 1 grams per liter. The “well-mixed” solution pours out at a rate of 5 liters/minute. Find the amount at time  $t$ .

We know

$$A'(t) = (\text{“flow - rate - in”})C_{in} - (\text{“flow - rate - out”})\frac{A(t)}{T(t)} = 4 - 5\frac{A(t)}{200 - t}, \quad A(0) = 30.$$

Writing this in the standard form  $A' + pA = q$ , we have

$$A(t) = \frac{\int \mu(t)q(t) dt + C}{\mu(t)},$$

where  $\mu = e^{\int p(t) dt} = e^{-5 \int \frac{1}{200-t} dt} = (200 - t)^{-5}$  is the “integrating factor”. This gives  $A(t) = 200 - t + C \cdot (200 - t)^5$ , where the initial condition implies  $C = -170 \cdot 200^{-5}$ .

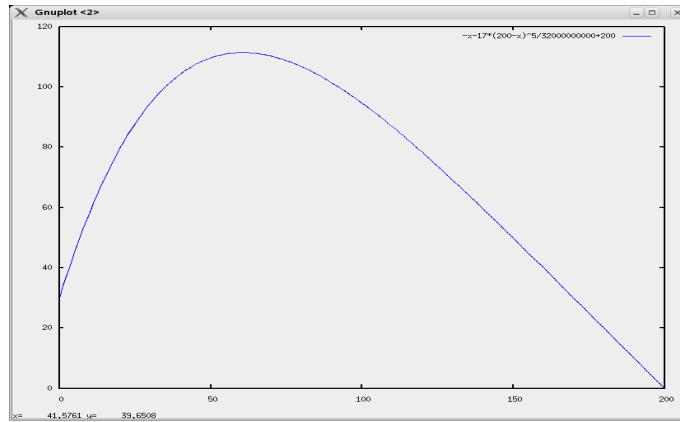


Figure 1:  $A(t)$ ,  $0 < t < 50$ ,  $A' = 4 - 5A(t)/(200 - t)$ ,  $A(0) = 30$ .

If you now solve the same problem but with the same flow rate out as 4 liters/min (so the “water level” in the tank is constant) then you get  $A(t) = 200 - 170e^{-t/50}$ , a much different function. This function looks like:

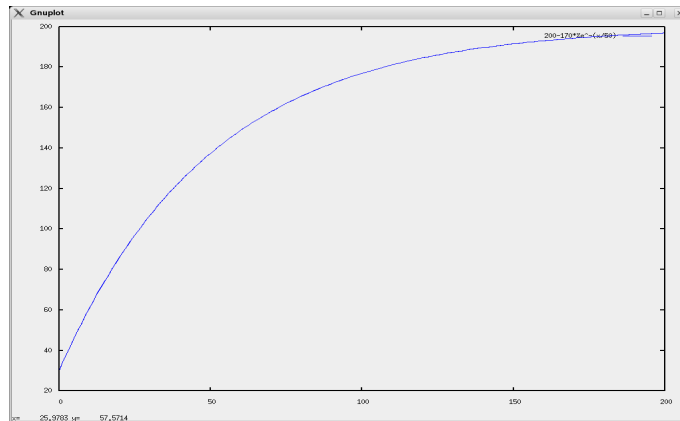


Figure 2:  $A(t)$ ,  $0 < t < 50$ ,  $A' = 4 - 4A(t)/200$ ,  $A(0) = 30$ .